

# Prediction of the Ultraviolet Luminescence Potential of $\text{Bi}_2\text{SeO}_5$ : First-Principles Insights

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**Abstract:** As a new layered semiconductor material,  $\text{Bi}_2\text{SeO}_5$  has shown potential in the field of ultraviolet electronic devices in recent years because of its unique crystal structure and wide band gap. In this paper, the crystal structure, electronic structure, and thermodynamic stability of  $\text{Bi}_2\text{SeO}_5$  are studied based on first-principles calculations. The ultraviolet luminescence property of  $\text{Bi}_{\text{Se}}$  defect is predicated from defect property, which provides theoretical basis for experimental design of high-performance  $\text{Bi}_2\text{SeO}_5$  photoelectric devices.

**Keywords:** First-principles calculation;  $\text{Bi}_2\text{SeO}_5$ ; Ultraviolet luminescence property

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## 1. Introduction

Traditional ultraviolet (UV) light sources are difficult to be portable, integrated, and automated due to their large size and power consumption. Compared with traditional UV light sources, nitride-based UV light sources have the advantages such as environmental friendliness, compact portability, ease of integration, tunable wavelengths, and long carrier lifetime. Therefore, nitride-based materials have great potential in UV luminescence applications. Although nitride UV sources have been widely used in secure communication, UV curing, optical medicine, and other fields, the low efficiency of luminescence is still an urgent problem to be solved, mainly due to the difficulty of p-type doping and the low carrier mobility. The preparation of high-quality materials and the realization of p-type doping are closely related to the defect properties of nitride. Effectively passivating defects to produce high-quality nitride materials persists as a significant technical hurdle.

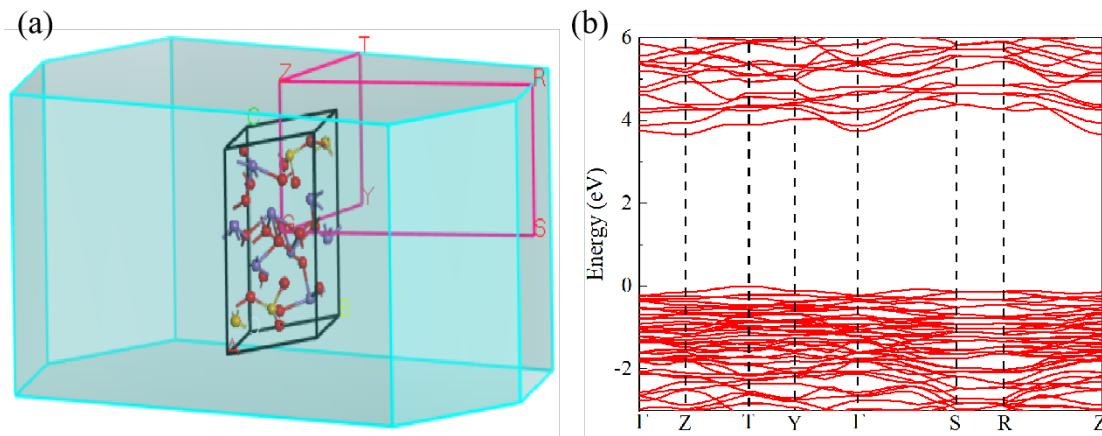
Semiconductors ( $\text{CsPbI}_3$ ,  $\text{Sb}_2\text{Se}_3$ ,  $\text{Sb}_2\text{S}_3$ ,  $\text{GeSe}$ , etc.) containing lone-pair electrons are usually defect-tolerant, that is, the defects with low formation energy have shallow defect levels, while the deep level defects have high formation energy<sup>[1-6]</sup>. Using  $\text{Bi}_2\text{SeO}_5$  as a gate dielectric layer, a high-performance field effect transistor can

be fabricated with low subthreshold swing, high current on/off ratios, and high mobility, which proves the high quality of the dielectric layer and few interface defects <sup>[7]</sup>.

However, little research has been done on  $\text{Bi}_2\text{SeO}_5$ . In 2001, Rademacher *et al.* successfully prepared  $\text{Bi}_2\text{SeO}_5$ . Unfortunately, they only described the crystal structure and did not study the material properties <sup>[8]</sup>. Subsequently, in 2004, Dityatyev *et al.* used neutron powder diffraction data to determine the crystal structure of  $\text{Bi}_2\text{SeO}_5$ , and studied its evolution with temperature <sup>[9]</sup>. In recent years, theoretical studies on  $\text{Bi}_2\text{SeO}_5$  have gradually increased <sup>[10–15]</sup>. In this paper, the crystal structure, electronic structure, and thermodynamic stability of  $\text{Bi}_2\text{SeO}_5$  are studied based on first-principles calculation. The ultraviolet luminescence of  $\text{Bi}_{\text{Se}}$  defect is predicted from the defect property, which provides a theoretical basis for experimental design of  $\text{Bi}_2\text{SeO}_5$  high-performance photoelectric devices.

## 2. Crystal and electronic structure

The primitive cell of  $\text{Bi}_2\text{SeO}_5$  contains 32 atoms, including 8 Bi atoms, 4 Se atoms, and 20 O atoms. Based on the symmetry of the crystal space group and the chemical environment of atomic bonding, there are three nonequivalent positions of Bi, six nonequivalent positions of O, and one nonequivalent position of Se in  $\text{Bi}_2\text{SeO}_5$ .



**Figure 1.** (a) The Brillouin zone and (b) band gap of  $\text{Bi}_2\text{SeO}_5$ .

In order to calculate the band gap of  $\text{Bi}_2\text{SeO}_5$ , we construct the Brillouin zone of  $\text{Bi}_2\text{SeO}_5$  through its crystal lattice characterization system. Based on the symmetry of the Brillouin zone, we then determine the high-symmetry paths and plan the required reciprocal space  $k$ -points for calculations. As shown in **Figure 1(a)**, the black line represents the crystal structure of  $\text{Bi}_2\text{SeO}_5$  with ABM space group, and its optimized lattice constants  $a = b = 8.77 \text{ \AA}$ ,  $c = 11.62 \text{ \AA}$ . The reciprocal space basis vectors are derived through the periodicity of the Bravais lattice and the Fourier transform of real-space basis vectors. The boundaries of the Brillouin zone are formed by the perpendicular bisecting planes of reciprocal lattice vectors, with the gray polygonal region enclosed by the blue lines representing the first Brillouin zone of  $\text{Bi}_2\text{SeO}_5$  in reciprocal space. Based on the symmetry of the Brillouin zone, the pink high-symmetry paths are extracted, and the high symmetry  $k$ -points ( $\Gamma$  (0,0,0), Z (0,0,0.5), T (-0.5, 0.5, 0.5), Y (-0.5, 0.5, 0), S (0, 0.5, 0) and R (0, 0.5, 0) that can completely reflect the topological characteristics of the energy band are selected. Connect successively adjacent high-symmetry points in the Brillouin zone to form a  $\Gamma \rightarrow Z \rightarrow T \rightarrow Y \rightarrow \Gamma \rightarrow S \rightarrow R \rightarrow Z$  closed path. On the high-symmetry path, symmetric points are sampled at equal

intervals using the uniform interpolation method. The 20 k-points are selected along the  $\Gamma$ -Z path, 28 k-points along the Z-T path, 20 k-points along the T-Y path, 28 k-points along the Y- $\Gamma$  path, 44 k-points along the  $\Gamma$ -S path, 20 k-points along the S-R path, and 44 k-points along the R-Z path. This results in a total of 204 k-points sampled across all high-symmetry paths. This k-point density can accurately capture band extremes while avoiding computational resource wastage caused by an excessive number of k-points.

Based on the first-principles calculation, we can plot the electronic band structure. First, the structure optimization of  $\text{Bi}_2\text{SeO}_5$  is performed to ensure both the lattice constants and the atoms' positions in equilibrium and stable state. In order to overcome the underestimation of the band gap by GGA, we employ the HSE hybrid function to correct the energy levels calculated by GGA. By mixing 18% Hartree-Fock exact exchange energy with 82% PBE exchange correlation energy, the accuracy of band gap calculation is effectively improved. The first principle calculations show that  $\text{Bi}_2\text{SeO}_5$  is an indirect bandgap semiconductor, the CBM is at Z point, the VBM is at T point, and the value of the band gap is 3.7 eV. The corresponding optical absorption edge of the wide band gap is 335 nm, indicating that  $\text{Bi}_2\text{SeO}_5$  can meet the UV luminescence requirements of shorter wavelengths. This approach provides theoretical guidance for designing UV  $\text{Bi}_2\text{SeO}_5$  light-emitting diodes (LEDs).

### 3. Thermodynamic stability

In order to synthesize single-phase  $\text{Bi}_2\text{SeO}_5$ , the chemical formulas of the constituent elements (Bi, Se, and O) must satisfy a series of thermodynamic conditions. First, the chemical potential should be satisfied.

$$2\mu_{\text{Bi}} + 5\mu_{\text{O}} + \mu_{\text{Se}} = \Delta H_f(\text{Bi}_2\text{SeO}_5)$$

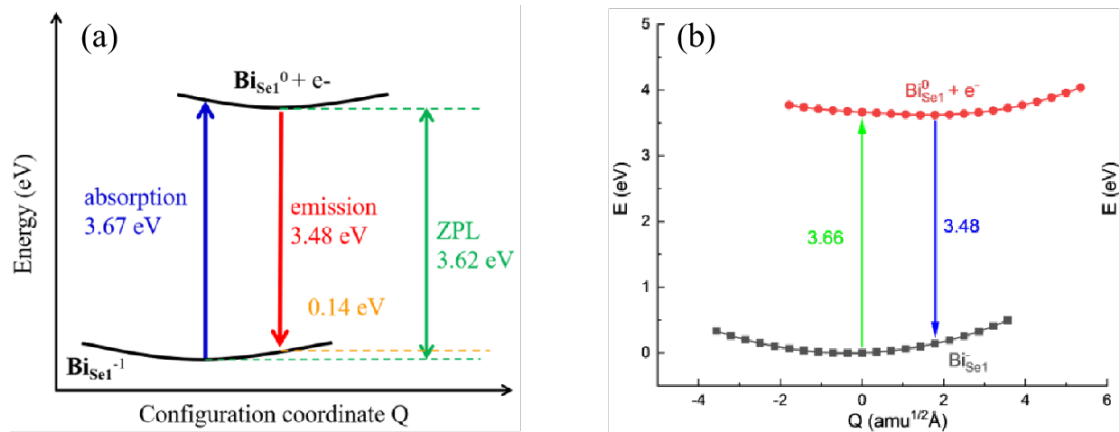
Here,  $\mu_i$  is the chemical potential of element  $i$  (Bi, Se, and O),  $\Delta H_f(\text{Bi}_2\text{SeO}_5)$  represents the formation enthalpy of  $\text{Bi}_2\text{SeO}_5$ . Second, to avoid the decomposition of  $\text{Bi}_2\text{SeO}_5$  into any Bi, O, or Se elemental, binary, and ternary compounds. For example, Bi,  $\text{O}_2$ , Se,  $\text{Bi}_2\text{O}_3$ ,  $\text{Bi}_2\text{Se}_3$ ,  $\text{BiO}_2$ ,  $\text{BiSe}_2$ ,  $\text{Se}_2\text{O}_5$ ,  $\text{SeO}_2$ ,  $\text{Bi}_2\text{O}_2\text{Se}$ ,  $\text{Bi}_2\text{Se}_3\text{O}_{10}$  and  $\text{Bi}_2\text{Se}_3\text{O}_9$  should also meet multiple inequality conditions, taking  $\text{Bi}_2\text{O}_3$  as an example:

$$2\mu_{\text{Bi}} + 3\mu_{\text{O}} < \Delta H_f(\text{Bi}_2\text{O}_3)$$

Based on the above restrictions, we can determine the range of elemental chemical potentials that need to be satisfied to form pure  $\text{Bi}_2\text{SeO}_5$ . As the chemical potential of O becomes poorer, the main competing phases of  $\text{Bi}_2\text{SeO}_5$  are  $\text{Bi}_2\text{O}_3$ . As the chemical potential of Se becomes richer,  $\text{Bi}_2\text{Se}_3$  and  $\text{Bi}_2\text{Se}_3\text{O}_9$  are the main competing phases. Therefore, the preparation of high-quality single crystal  $\text{Bi}_2\text{SeO}_5$  requires stricter control of  $\mu_{\text{Se}}$ .

### 4. Defect-induced photoluminescence

The photoluminescence mechanism induced by defects in semiconductors involves the process where charge carriers are photoexcited to an excited state and subsequently return to the ground state through radiation recombination, releasing photons. This paper adopts configuration coordinate diagrams combined with a one-dimensional vibrational approximation to investigate the UV luminescence property of  $\text{Bi}_2\text{SeO}_5$ . For the radiation transition, the configuration coordinate diagram of  $\text{Bi}_{\text{Se1}}$  defect is drawn, which shows the excitation and emission process of the charge carrier.



**Figure 2.** (a) Configuration coordinate diagram of the photoluminescence in BiSe defect and (b) a harmonic vibrational potential energy surface of BiSe defect luminescence process.

For the  $\text{Bi}_{\text{Se}}$  defect under the condition of p-type conduction, it exhibits a tendency to accept electrons, which is an acceptor, resulting in a relatively high concentration of  $\text{Bi}_{\text{Se}}^-$ . The  $\text{Bi}_{\text{Se}}^-$  defect state is the ground state, the photoexcitation process causes the valence state of  $\text{Bi}_{\text{Se}}$  defect to transition from the ground state  $\text{Bi}_{\text{Se}}^-$  to the excited state  $\text{Bi}_{\text{Se}}^0$  (the electron on the  $\text{Bi}_{\text{Se}}^-$  defect level is excited to the CBM, generating an electron at CBM). The electron in the  $\text{Bi}_{\text{Se}}^0$  excited state is relaxed, then transitions back to the  $\text{Bi}_{\text{Se}}^-$  ground state to generate a photon. For the electron trapping process of the defect, the energy of the zero phonon line is the difference value between the band gap and the transition level. As shown in **Figure 2(a)**, the relaxation energy between the -1 and neutral valence states of  $\text{Bi}_{\text{Se}}$  is very small (only 0.14 eV). When excited by the light with energy of 3.67 eV,  $\text{Bi}_{\text{Se}}$  can emit UV light with energy of 3.48 eV, indicating a potential UV light source. The transition between the ground and excited states of  $\text{Bi}_{\text{Se}}$  occurs on a harmonic vibrational potential energy surface, as depicted in **Figure 2(b)**. The minimal shift in the normal coordinate  $\Delta Q$  signifies an extremely small structural distortion induced by photoexciting in  $\text{Bi}_{\text{Se}}$  defect, further supporting its viability as a potential UV light source.

## 5. Conclusion

As a new layered semiconductor material,  $\text{Bi}_2\text{SeO}_5$  has shown great potential in the field of UV optoelectronic devices due to its unique crystal structure and wide band-gap. In this paper, we systematically analyze the physical basis and technical advantages of the ultraviolet luminescence performance of  $\text{Bi}_2\text{SeO}_5$  from its crystal structure, thermodynamic stability, electronic structure, and luminescence mechanism of  $\text{Bi}_{\text{Se}}$  defect. The first-principles theoretical calculation provides a theoretical basis for the experimental design of high-performance  $\text{Bi}_2\text{SeO}_5$  optoelectronic devices.

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## Disclosure statement

The authors declare no conflict of interest.

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